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S-16 Chapter 2 Water In a typical analysis, 15 mL of an aqueous solution containing an unknown amount of acetylcholine had a pH of 7.65 When incubated with acetylcholinesterase, the pH of the solution decreased to 6.87

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Water - Students Manuals

solution of bicarbonate (NaHCO_3) its OOH group ionizes, making the compound much more water-soluble A neutral or acidic solution, as in (a) or (b), would not be effective 15 pH and Drug Absorption Aspirin is a weak acid with a pK_a of 3.5 (the ionizable H is shown in bold):

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Protein solution is added and electric field is reapplied After staining, proteins are shown to be distributed along pH gradient according to their pI values A stable pH gradient is established in the gel after application of an electric field An ampholyte solution is pH 9 incorporated into a gel

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